## Synthesis of gluco-Configured Tetrahydroimidazopyridine-2 phosphonate-Derived Lipids, Potential Glucosyl Transferase Inhibitors

by Isabelle Billault and Andrea Vasella\*

Laboratorium für Organische Chemie, ETH-Zentrum, Universitätstrasse 16, CH-8092 Zürich

The analogues  $1 - 3$  of dolichol monophosphatidyl  $\beta$ -p-glucose have been prepared as potential inhibitors of the glucosyl transferase Alg10p. Pd(PPh<sub>3</sub>)<sub>4</sub>-catalysed phosphonylation of the iodoimidazole 4 with diethyl, dimethyl, and diphenyl phosphite led to the corresponding phosphonic acid diesters, which were transformed into deprotected and silyl-protected diesters, deprotected monoesters, and protected and unprotected phosphonic acids (Scheme). A N-methyl imidazolium salt was obtained as a by-product of the dimethylphosphonylation of the iodoimidazole, and prepared in high yields by methylation of the imidazole 8 with MeI; the corresponding deprotected salt 11 inhibits sweet almond  $\beta$ -glucosidases (IC<sub>50</sub> = 308 µm). Trichloroacetonitrile-promoted monoesterification of the acetylated mono-triethylammonium salt 19 with oleyl alcohol, phytanol, and dolichol-19, followed by deacetylation, gave the desired glycophospholipids.

Introduction. - Glycosyl transferases catalyse the regio- and stereoselective formation of a glycosidic bond between the reducing end of a mono- or oligosaccharide and a defined heteroatom (O or N) of their acceptor substrate<sup>1</sup>). The glycosyl donor is either a monosaccharide activated as a nucleoside diphosphate, a nucleoside monophosphate, or a dolichyl monophosphate, or then an oligosaccharide activated as a dolichyl monophosphate. Most glycosyl transferases transfer the glycosyl moiety with inversion of configuration at the anomeric centre of the transferred sugar residue [1]. There is a great amount of amino-acid sequence data, available on the internet<sup>2</sup>), that have been classified into sequence-related families [1]. However, crystal-structure details of glycosyl transferases are very scarce [2] [3], and information about the active site and the amino-acid residues directly involved in the catalysis is limited  $[4-6]$ . Analysis of the isotope effects [7] [8] and inhibition studies [9] [10] strongly suggest that the reactive intermediate resembles a glycosyl cation similar to the reactive intermediate of the enzymic glycoside hydrolysis. It is, however, not clear whether the phosphate leaving group is protonated at one of its O-centres (by a functional equivalent of the catalytic acid in glycosidases) or activated by coordination to a metal ion  $(Mg^{2+}, Mn^{2+})$ , or whether a basic amino-acid residue, accepting a proton from the hydroxy or amido function of the acceptor is implied in catalysis. fuco-Configured azonia sugars, mimicking the positive charge of the putative cationic intermediate, inhibit fucosidases strongly, but fucosyl transferases only weakly. Addition of GDP

<sup>&</sup>lt;sup>1</sup>) The acceptor of a monosaccharide is either a mono- or oligosaccharide or a phospholipid, and the acceptor of an oligosaccharide is either a peptide or a protein.

<sup>&</sup>lt;sup>2</sup>) B. Henrissat and P. Coutinho at URL http://afmb.cnrs-mrs.fr/ $\sim$  pedro/CAZY/db.html.

 $(guanosine 5'-diphosphate)$  improves the inhibition of fucosyl transferases by 'azonia sugars<sup>3</sup>) [9] [11] [12], and the nucleotide part of the fucosyl donor, *viz*. GDP, acts itself as inhibitor of fucosyl transferases. Analogues of UDP-Gal and GDP-Fuc with a nucleotide moiety bound to an unsaturated glycosyl or carba-glycosyl residue derived from glycosidase inhibitors mimic the shape of the hypothetical intermediate, and also inhibit galactosyl and fucosyl transferases, respectively [13] [14]. Thus, potential inhibitors of glycosyl transferases may be obtained by appropriate modifications of glycosidase inhibitors  $[11 - 14]$ , in spite of the obvious differences between glycosidases and glycosyl transferases.

We became interested in the inhibition of the glucosyl transferase Alg10p involved in the N-glycosylation process [15]. Alg10p is a transmembrane enzyme localized in the endoplasmic reticulum (ER). It uses  $\beta$ -D-Glc dolichyl-monophosphate [16] [17] as glucosyl donor and catalyses the formation of a  $Glc\alpha(1,2)Glc$  bond [18]. Neither the exact role of the dolichyl monophosphate moiety, nor the specificity of Alg10p are known. Inhibitors of Alg10p have not yet been reported.

The strong inhibition of retaining  $\beta$ -glycosidases by gluco-, manno-, and galactoconfigured tetrahydroimidazopyridines  $[19-21]$  has been attributed mainly to the combined effect of a (partial) protonation of the imidazole moiety by the catalytic acid and an electrostatic interaction between the imidazolium cation and the catalytic base [22]. As Alg10p uses a  $\beta$ -d-configured glucosyl donor, it may interact similarly with a gluco-configured tetrahydroimidazopyridine. We planned to prepare the glucophospholipid analogues  $1-3$  from the known benzyl-protected imidazole 8 [19] via the phosphonates  $5 - 7$  (*Scheme*), introducing oleyl, phytanyl, and dolichyl substituents to evaluate the selectivity of Alg10p for the lipid moiety.



**Synthesis.**  $-$  The required imidazolylphosphonates  $5-7$  were obtained *via* the mono-iodoimidazole 4 that was prepared [23] similarly to the known bromo analogue [21] [24] (Scheme). The phosphonate group was introduced by  $Pd(PPh<sub>3</sub>)<sub>4</sub>$ -catalysed cross-coupling with dialkyl or diphenyl hydrogen phosphites  $[25 - 29]$ . Treatment of 4 with diethyl hydrogen phosphite in the presence of  $Et_3N$  and Pd(PPh<sub>3</sub>)<sub>4</sub> led to a mixture of the diethyl phosphonate 5 and the unsubstituted imidazole 8. A  $\frac{5}{8}$  ratio of ca. 7:3 was determined from the integrals of the benzyl <sup>1</sup>H-NMR signals at 5.10 (5) and

<sup>&</sup>lt;sup>3</sup>) The synergism of the inhibition of GDP and 'azonia sugars' is expressed by a  $22-77$ -fold decrease of the  $IC_{50}$  for the piperidinium salts upon addition of GDP [11].

5.19 ppm  $(8)$ ; the phosphonate 5 was isolated in 62% yield<sup>4</sup>). The ratio 5/8 depends strongly on the concentration of the iodoimidazole 4 and the amount of the catalyst. In the presence of 0.3 equiv. of  $[Pd(PPh_3)_4]$ , the ratio 5/8 ranged from 1:1 (0.15m 4) to 7:3  $(0.3M<sub>4</sub>)$ , and reached 9:1 in the presence of 1 equiv. [Pd(PPh<sub>3</sub>)<sub>4</sub>]  $(0.3M<sub>4</sub>)$ . To suppress the formation of 8, we tested a range of amines and phosphites (*Table 1*). The bulkiest base, 1,2,2,6,6-pentamethylpiperidine (PMP), led to the highest 5/8 ratio and improved the yield of  $5$  to  $71\%$  (*Entry 3*).



a)  $[Pd(PPh_3)_4]$ , Et<sub>3</sub>N, toluene, HPO(OR)<sub>2</sub>; 5 (62%); 6 (30%); 7 (84%). b) MeI, toluene, 95°; 98%. c) 20%  $Pd(OH)_2/C$ , AcOH, AcOEt/MeOH/H<sub>2</sub>O, H<sub>2</sub>; 89%. d) 'BuMe<sub>2</sub>SiCl, 1H-imidazole, 25°, 48 h. e) aq. NaOH soln., 70°, Amberlite IRC 50 (H<sup>+</sup> form); 58%.  $f$ ) 1m aq. HCl soln., Bio-Rad AG 2-X8 resin (Cl<sup>-</sup> form). g) 20% Pd(OH)<sub>2</sub>/C, AcOH, MeOH/H<sub>2</sub>O, H<sub>2</sub>; 90%. h) Me<sub>3</sub>SiBr, CH<sub>2</sub>Cl<sub>2</sub>. i) Pd(OH)<sub>2</sub>, MeOH/AcOEt/H<sub>2</sub>O 3:1:1, H<sub>2</sub>; 81% (from 5). j) Ac<sub>2</sub>O, pyridine. k) Dowex 50W8 (Na<sup>+</sup> form), Dowex CCR-2 (Na<sup>+</sup> form). l) CCl<sub>3</sub>CN, pyridine, oleyl alcohol for 20 (50% from oleyl alcohol), phytanol for 21 (61% from phytanol), or dolichol-19 for 22 (65% from dolichol-19). m) MeONa, MeOH, *Dowex IRC* 50 (H<sup>+</sup> form); 1 (95%); 2 (66%); 3 (75%).

<sup>4)</sup> Excess diethyl hydrogen phosphite led to increased amounts of triphenylphosphine oxide that had to be removed by repeated chromatography.

Entry	Phosphite	$Basea$ )	Time (h)	$Ratiob$ ) 5/8	6/8	7/8	Yield $[\%]$ 5	6	7
1	$HPO(OEt)$ ,	Et <sub>3</sub> N	26	70:30			62		
2	$HPO(OEt)$ ,	$(iPr)$ <sub>2</sub> EtN	18	83:17			60		
3	HPO(OEt),	<b>PMP</b>	19.5	95:05			71		
$4^{\circ}$ )	$HPO(OEt)$ <sub>2</sub>	Et <sub>3</sub> N	21	50:50					
5	HPO(OME),	Et <sub>3</sub> N	19		74:26			30	
6	HPO(OME),	$(^{i}Pr)_{2}EtN$	20		d١			20	
7	HPO(OME),	<b>PMP</b>	19.5		d			10	
8	$HPO(OPh)$ ,	Et <sub>3</sub> N	19			100:0			84
9	$HPO(OPh)$ ,	$(iPr)$ <sub>2</sub> EtN	18			100:0			83
10	$HPO(OPh)$ ,	<b>PMP</b>	19.5			100:0			78
$11^{\circ}$	$HPO(OPh)$ ,	Et <sub>3</sub> N	17			85:15			74

Table 1. Phosphonylation of 4 (0.4m 4 in toluene at 95° with 5 equiv. of HPO(OR), 7 equiv. of base, and 0.3 equiv. of  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$ 

<sup>a</sup>) PMP: 1,2,2,6,6 pentamethylpiperidine. <sup>b</sup>) The ratio was determined on the basis of integrals of the benzyl <sup>1</sup>H-NMR signals. <sup>c</sup>) Reaction performed in the presence of CuI (0.36 equiv.). <sup>d</sup>) The ratio could not be determined, because of the insufficient amount of the expected products (6/8).

While the Pd<sup>0</sup>-catalysed phosphonylation of a bromoimidazole with dimethyl hydrogen phosphite failed [29], dimethyl-phosphonylation of the iodoimidazole 4 in the presence of  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$  and Et<sub>3</sub>N yielded 30% of 6 (*Entry 5*). Using ( ${}^{\text{P}}P$ <sub>2</sub>EtN instead of  $Et_3N$  (*Entry 6*) proved detrimental. The major by-product of this coupling was an N-methylimidazolium salt that was isolated by flash chromatography (silica gel). Its NMR spectra are very similar to those of the iodide 10, obtained by treating 8 with MeI in toluene (4 h at  $95^{\circ}$ ). Presumably, 10 and the by-product differ only by the nature of the counterion. The unprotected N-methylimidazolium salt 11 was prepared in almost quantitative yields by methylation of 9. With an  $IC_{50}$  value of 308  $\mu$ m (37°, pH 6.8, phosphate buffer), 11 is a rather weak inhibitor of sweet-almond  $\beta$ -glucosidases [30].

Coupling of the iodide 4 with diphenyl hydrogen phosphite yielded  $78 - 84\%$  of 7 (*Entries*  $8-10$ ) without forming any of the dehalogenated imidazole **8**. To the best of our knowledge, this is the first example of a  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$ -catalysed coupling of diphenyl hydrogen phosphite with a halo(het)arene. Coupling 4 with either diethyl or diphenyl hydrogen phosphite in the presence of both  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$  and CuI, as in the *Sonogashira* reaction [31] [32], lowered the ratio  $5/8$  and  $7/8$  (*Entries 4* and 11, resp.).

We briefly studied the deprotection of the phosphonates and the introduction of alternative protecting groups. Catalytic hydrogenolysis of the diethyl phosphonate 5 led cleanly to the tetrol 12. Silylation of 12 yielded 87% of 13. Saponification of 12 gave the monoester 14, which was transformed into its hydrochloride, characterized by a p $K_{HA}$  of 5.15. Hydrogenolysis of 7 provided the monophenyl ester 15 in excellent yields.

The synthesis of  $1 - 3$  was continued by dealkylating the diethyl phosphonate 5 with  $Me<sub>3</sub>SiBr$  [33] to the phosphonic acid 16 which was subjected to hydrogenolytic debenzylation. The product was purified by chromatography on DEAE-cellulose (elution with aq.  $HEt_3N^+HCO_3^-$ ) to afford a mixture of mostly the mono(triethylammonium) salt 17 and varying amounts of the corresponding acid  $(81\%; 0.70 - 0.95)$  equiv.

of  $Et_3N$  by <sup>1</sup>H-NMR). Alternatively, 17 was obtained by treating the hydrogenolysis product with  $Et<sub>3</sub>N$ , followed by lyophilization; this preparation also contained between 0.70 and 0.95 equiv. of Et<sub>3</sub>N. Treatment of 17 with *Dowex* 50W8 (Na<sup>+</sup> form) and then *Dowex CCR-2* (Na<sup>+</sup> form) led to the disodium salt **18**. The  $pK_{HA}$  values of **18** (7.69 and 4.84) were determined by titration of an aqueous solution with 0.1 HCl at  $25^{\circ}$ . The higher  $pK_{HA}$ , corresponding to the dissociation constant of the phosphonate group, is close to the p $K_{HA}$  value of pyridin-2-ylphosphonic acid (p $K_a = 7.71$ ) [34]. The second  $pK_{H_A}$  of 18 is slightly lower than that of the *gluco*-tetrahydroimidazopyridine 9 ( $pK_{HA} = 6.10$ ). The third  $pK_{HA}$  value was not determined. Acetylation of 17 (Ac<sub>2</sub>O, pyridine) gave the tetraacetate 19. As chromatography of 19 (DEAE-cellulose) led to partial deacetylation, it was used without further purification and esterified with 0.87 equiv. of oleyl alcohol  $(=(Z)$ -octadec-9-en-1-ol) using trichloroacetonitrile in pyridine as coupling agent [35] [36] to yield 50% of the oleyl phosphonate 20. Other coupling agents, such as bromotris(dimethylamino) phosphonium hexafluorophosphate (BroP ) [37] [38], 2,4,6-triisopropylbenzensulfonyl chloride [39], oxalyl chloride, DMF (cat.) [40], and diethyl diazenedicarboxylate (DEAD)/PPh<sub>3</sub> [41] [42] either led to incomplete transformation of the alcohol or to byproducts. Esterification of 19 with 0.80 equiv. of phytanol yielded 61% of the phytanyl phosphonate 21. Similarly, esterification with dolichol-19 ( $= 3,7,11,15,19,23,27,31,35,39,43,47,51,55,59,63,67,71,75-$ 

nonadecamethylhexaheptaconta-6,10,14,18,22,26,30,34,38,42,46,50,54,58,62,66,70,74-octadecaen-1-ol), but using 5 equiv. of the phosphonate  $19<sup>5</sup>$ , provided 65% of the dolichyl phosphonate 22. Deacetylation of  $20 - 22$  (NaOMe in MeOH followed by *Amberlite* IRC 50 (H<sup>+</sup> form)) led to 1, 2, and 3 in 95%, 66%, and 75% yield, respectively.

This synthesis provides the gluco-configured tetrahydroimidazopyridine-2-phosphonates  $1 - 3$  in six steps from the iodoimidazole 4 and in overall yields of 24, 20 and 24%, respectively. The iodoimidazole 4 is available in five steps and 65% overall yield from the readily available  $2,3,4,6$ -tetra-O-benzyl-D-gluconolactam [19] [43]. The inhibitory effect of 1–3 on the  $(a1 \rightarrow 2)$  glucosyl transferase Alg10p is under investigation.

Formation of the C(2)–P bond in  $5-7$  is evidenced by <sup>1</sup>J(C(2),P) of 247.4 and 256.4 Hz in the <sup>13</sup>C-NMR spectra of 6 and 7, respectively. Signal overlap did not allow the determination of  $^1J(C(2),P)$  of 5. However, the <sup>13</sup>C-NMR spectra of 17 and 18, derived from 5 show a  $\frac{1}{J}(C(2), P)$  of 190.4 and 202.6 Hz, respectively. The conformation of the phosphonate 17 in  $D_2O(TH_6)$  is very similar to the one of the imidazole 9 [19] (Table 2). The FAB-MS of  $20 - 22$  evidence the formation of a monoester in each case. This is further corroborated by the shift of the <sup>31</sup>P-NMR signal from  $-1.12$  ppm (CD<sub>3</sub>OD) for 19 to 3.98 ppm (CD<sub>3</sub>OD) for 20, 3.78 ppm (CD<sub>3</sub>OD) for 21, and 3.87 ppm (CD<sub>3</sub>OD/CDCl<sub>3</sub>4:2) for 22. The CH<sub>2</sub>O signal of the alkoxy moiety of 20 - 22 is also shifted to characteristically lower field. The phosphonates 1 and 2, but not 3, are sufficiently well soluble in  $CD<sub>3</sub>OD$  to lead to sharp  $^1$ H-,  $^{13}$ C-, and  $^{31}$ P-NMR signals, while the spectra of  $3$  (CDCl<sub>3</sub>/CD<sub>3</sub>OD/D<sub>2</sub>O 10 : 7 : 1) showed only broad signals. The conformation of the saccharide moiety of  $1-3$  should be similar to that of 9 and 17  $(7H<sub>6</sub>)$ . This is fully evidenced only for 1 for which all coupling constants could be determined. The signals of H $-C(5)$  or H $-C(6)$  of 2 are hidden, but  $J(7,8)$  and  $J(6,7)$  are in agreement with the expected conformation (Table 2).

<sup>5)</sup> The phosphonate 19 was used in excess, considering the price of dolichol-19.

	9	17	18		
Solvent	D <sub>2</sub> O	D <sub>2</sub> O	D <sub>2</sub> O	CD <sub>3</sub> OD	CD <sub>3</sub> OD
J(8,7)	8.7	9.7	9.0	8.0	8.1
J(7,8)	9.7	9.7	10.0	9.5	8.8
J(6,5)	9.7	8.8	10.0	8.0	a <sub>1</sub>
$J(5,CH-C(5))$	2.5	3.1	2.5	4.0	4.1
$J(5,CH'-C(5))$	2.2	2.2	2.5	2.0	a
${}^{2}J(CH, -C(5))$	12.8	13.2	13.0	11.8	13.5

Table 2. Coupling Constants J [Hz] of Tetrahydroimidazopyridine 9 and Tetrahydroimidazopyridine-2-phosphonates 1, 2, 17, and 18

We thank T. Mäder for the HPLC purifications, Dr. B. Bernet for checking the experimental part, M. Schneider and D. Manser for the determination of the  $pK_{H\text{A}}$  values, and the Swiss National Science Foundation and F. Hoffmann-La Roche AG, Basel, for generous support.

## Experimental Part

General. Solvents were distilled before use, toluene was degassed, and reactions were run under Ar.  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$  (*Aldrich*) and CuI (*Fluka*) were used without purification. Oleyl alcohol (tech. 85%, *Aldrich*) was purified by FC (hexane/AcOEt 9:1) followed by FC (RP-18 silica gel; MeOH/H<sub>2</sub>O 9:1  $\rightarrow$  85:15). Phytanol was obtained in 64% yield (following the procedure of Sakata et al. [44]) by catalytic hydrogenation of phytol (Fluka) in the presence of Pd/C (10%) for 2 h. Dolichol-19 was purchased from the Polish Academy of Sciences ( Institute of Biochemistry and Biophysics) and purified by FC (toluene) before use. TLC: Merck silica gel  $60F_{254}$  plates; detection by heating with 'mostain' (400 ml of 10% H<sub>2</sub>SO<sub>4</sub> soln., 20 g of (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub> · 6 H<sub>2</sub>O, 0.4 g of  $Ce(SO_4)$ ). Flash chromatography (FC): silica gel 60 (Fluka; 0.04–0.063 mm), unless indicated otherwise. M.p.: uncorrected.  $^1H$ -,  $^{13}C$ -, and  $^{31}P$ -NMR Spectra: chemical shifts  $\delta$  in ppm rel. to TMS ( $^1H$  and  $^{13}C$ ) or  $H_3PO_4$  ( $^{31}P$ ) as external standard, and coupling constants J in Hz. FAB-MS: 3-nitrobenzyl alcohol as matrix, unless indicated otherwise.

Diethyl (5R,6R,7S,8S)-6,7,8-Tris(benzyloxy)-5-[(benzyloxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2-a]pyridine-2-phosphonate (5). A mixture of 4 (1.0 g, 1.45 mmol) and  $[Pd(PPh<sub>3</sub>)<sub>4</sub>]$  (505 mg, 0.44 mmol) in toluene  $(3.6 \text{ ml})$  was treated with Et<sub>3</sub>N (1.4 ml, 10 mmol) and HPO(OEt)<sub>2</sub> (0.94 ml, 7.28 mmol), warmed to 95°, and stirred for 19 h. After the addition of AcOEt (10 ml), the suspension was filtered through Celite and the residue washed with AcOEt (300 ml). The filtrate was concentrated to 150 ml, washed (H<sub>2</sub>O), dried (MgSO<sub>4</sub>), filtered, and evaporated. <sup>1</sup>H-NMR of the crude showed a mixture  $5/8$  ca. 70:30, besides P(O)Ph<sub>3</sub> and HPO(OEt)<sub>2</sub>. FC (hexane/AcOEt/Et<sub>3</sub>N 7:3:0.03  $\rightarrow$  0:1:1:0.03) gave 8 [19] (242 mg, 30%) and a brown mixture containing principally 5 and P(O)Ph<sub>3</sub> (717 mg). FC of this mixture (CH<sub>2</sub>Cl<sub>2</sub>/i-PrOH 10 : 0.05  $\rightarrow$  10 : 0.5) gave a colourless oil  $5/P(O)Ph_3 ca. 80 : 20. PO(Ph)_3$  was removed by FC (RP-C18 silica gel; MeOH/H<sub>2</sub>O 8 :  $2 \rightarrow 9 : 1$ ): 5 (624 mg, 62%) as a colourless oil.  $R_f$  (hexane/AcOEt/Et<sub>3</sub>N 1:0.03) 0.2. UV (CHCl<sub>3</sub>): 269 (2.81). IR (CCl<sub>4</sub>): 3065w, 3032w, 2980w, 2929w, 2906w, 2867w, 1497w, 1454m, 1438w, 1362w, 1264m, 1235m, 1203m, 1117s, 1098s, 1062s, 1030s, 968m. <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz): 1.35 (br.  $q$ ,  $J = 6.4$ , 2 Me); 3.76 (dd,  $J = 5.2$ , 10.2, CH $-C(5)$ ); 3.81 – 3.90 (m, irrad. at 4.11  $\rightarrow$  change, CH'–C(5), H–C(6)); 4.12 (dd, J = 5.1, 6.7, irrad. at 3.84  $\rightarrow$  change, H–C(7));  $4.06 - 4.30$  (m, 5 H, irrad. at  $1.35 \rightarrow$  change, 2 MeCH<sub>2</sub>O, irrad. at  $3.84 \rightarrow$  change, H $-C(5)$ ); 4.46 (br. s, PhCH<sub>2</sub>); 4.47 (d, J = 11.8, PhCH); 4.64 (d, J = 11.3, PhCH); 4.77 (d, J = 5.1, irrad. at 4.12  $\rightarrow$  change, H  $-C(8)$ ); 4.78  $(d, J = 11.8, PhCH)$ ; 4.80  $(d, J = 11.7, PhCH)$ ; 4.82  $(d, J, 11.3, PhCH)$ ; 5.10  $(d, J = 11.7, PhCH)$ ; 7.12 - 7.22  $(m, 2 \text{ arom. H})$ ; 7.23 – 7.42  $(m, 18 \text{ arom. H})$ ; 7.71  $(s, H - C(3))$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 75 MHz): 16.11  $(dq, {}^{3}J(C, P) = 5.9,$  Me); 16.15  $(dq, {}^{3}J(C, P) = 5.9,$  Me); 58.15  $(d, C(5))$ ; 62.23  $(t, {}^{2}J(C, P) = 4.9,$  2 CH<sub>2</sub>O); 67.71  $(t, CH<sub>2</sub> - C(5))$ ; 72.23, 73.26, 73.57, 73.88 (4t, 4 PhCH<sub>2</sub>); 73.26, 75.88, 81.46 (3d, C(6), C(7), C(8)); 127.42  $(dd, {}^2J(C, P) = 40.3, C(3))$ ; 127.68 – 128.54 (several d); ca. 131.87  $(d, {}^1J(C, P) \approx 250, C(2))$ ; 137.08, 137.41, 137.60, 137.97 (4s); 146.55  $(d, {}^{3}J(C, P) = 22.0, C(8a))$ . <sup>31</sup>P-NMR (CDCl<sub>3</sub>, 121 MHz): 12.21. FAB-MS: 697 (100, [*M* +  $1$ ]<sup>+</sup>), 1393 (20, [2*M* + 1]<sup>+</sup>). Anal. calc. for C<sub>40</sub>H<sub>45</sub>N<sub>2</sub>O<sub>7</sub>P (696.78): C 68.95, H 6.51, N 4.02; found: C 68.98, H 6.66, N 4.21.

Dimethyl (5R,6R,7S,8S)-6,7,8-Tris(benzyloxy)-5-[(benzyloxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2-a]pyridine-2-phosphonate (6). A mixture of 4 (20 mg, 29.15  $\mu$ mol) and [Pd(PPh<sub>3</sub>)<sub>4</sub>] (10 mg, 8.74  $\mu$ mol) in toluene (73  $\mu$ ) was treated with Et<sub>3</sub>N (28  $\mu$ , 0.2 mmol) and HPO(OMe)<sub>2</sub> (13  $\mu$ , 0.14 mmol), warmed to 95°, and stirred for 19 h. The mixture was concentrated and co-evaporated with toluene. The <sup>1</sup>H-NMR spectrum of the crude showed a mixture 6/8 ca. 70:30, besides P(O)Ph<sub>3</sub> and HPO(OMe)<sub>2</sub>. FC (hexane/AcOEt  $5:5 \rightarrow 0:1$ ) gave 8  $(3 \text{ mg}, 18\%)$  and 6 as coloured oils. FC on RP-C18 silica gel (MeOH/H<sub>2</sub>O 8 : 2) gave 6 (6 mg, 31%). Colourless oil. R<sup>f</sup> (AcOEt) 0.26. IR (CCl4) 3089w, 3065w, 3032w, 2950w, 2929w, 2854w, 1515w, 1497w, 1454m, 1361w, 1334w, 1268m, 1239m, 1207w, 1185w, 1098s, 1064s, 1039s. <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz): 3.75 (dd, J = 5.3, 10.3, CH – C(5)); 3.78 – 3.89 (*m*, irrad. at 4.11 or 4.23  $\rightarrow$  change, CH' – C(5), H – C(6)); 3.80 (*d*, <sup>3</sup>*J*(H,P) = 7.0, Me); 3.84  $(d, {}^{3}J(H, P) = 7.0, Me)$ ; 4.11  $(dd, J = 4.8, 6.9, irrad.$  at  $4.76 \rightarrow d, J \approx 6.8, H - C(7))$ ; 4.23  $(dd, J = 2.7, 4.8, 7.6,$  $H-C(5)$ ; 4.46 (d, J = 11.4, PhCH); 4.46 (t, J = 13.0, PhCH<sub>2</sub>); 4.62 (d, J = 11.5, PhCH); 4.76 (d, J = 4.9, irrad. at  $4.11 \rightarrow s$ , H $-C(8)$ ); 4.77 (d, J = 11.4, PhCH); 4.78 (d, J = 11.5, PhCH); 4.80 (d, J = 11.8, PhCH); 5.08 (d,  $J = 11.5$ , PhCH); 7.12 – 7.20 (m, 2 arom. H); 7.22 – 7.42 (m, 18 arom. H); 7.72 (s, H – C(3)). <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 75 MHz): 52.87  $(dq, {}^{2}J(C, P) = 6.4,$  Me); 53.00  $(dq, {}^{2}J(C, P) = 6.4,$  Me); 58.33  $(d, C(5))$ ; 67.83  $(t, CH_{2}-C(5))$ ; 72.40, 73.35, 73.63, 73.92 (4t, PhCH<sub>2</sub>); 73.35, 75.92, 81.38 (3d, C(6), C(7), C(8)); 127.79 - 128.64 (several d); 129.01  $(d, {}^{1}J(C, P) = 247.4, C(2))$ ; 137.12, 137.44, 137.63, 138.01  $(4s)$ ; 146.77  $(d, {}^{3}J(C, P) = 22.3, C(8a))$ . <sup>31</sup>P-NMR  $(CDCl_3, 121 MHz)$ : 14.84. FAB-MS: 669 (100,  $[M+1]^+$ ), 1338 (10,  $[2 (M+1)]^+$ ).

Diphenyl (5R,6R,7S,8S)-6,7,8-Tris(benzyloxy)-5-[(benzyloxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2-a]pyridine-2-phosphonate (7). A mixture of 4 (20 mg, 29.15  $\mu$ mol) and  $[Pd(PPh_3)_4]$  (10 mg, 8.74  $\mu$ mol) in toluene (73  $\mu$ ) was treated with Et<sub>3</sub>N (28  $\mu$ , 0.20 mmol) and HPO(OPh)<sub>2</sub> (28  $\mu$ , 0.14 mmol), warmed to 95°, and stirred for 19 h. The mixture was concentrated and co-evaporated with toluene. The <sup>1</sup>H-NMR spectrum of the crude showed 7, besides P(O)Ph<sub>3</sub> and HPO(OPh)<sub>2</sub>. FC (hexane/AcOEt  $8:2 \rightarrow 7:3 \rightarrow 6:4$ ) followed by FC (CH<sub>2</sub>Cl<sub>2</sub>/i-PrOH 10: 0.05) gave 7 (19.4 mg, 84%). Colourless oil.  $R_f$  (hexane/AcOEt 4: 6) 0.47. IR (CCl<sub>4</sub>): 3066w, 3032w, 2923w, 2865w, 1593m, 1490s, 1454m, 1361w, 1280m, 1243w, 1216m, 1193s, 1162m, 1099m, 1070m, 1006w, 986w, 941s. <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 300 MHz): 3.71 (dd, J = 5.4, 10.7, irrad. at  $4.19 \rightarrow d, J \approx 10.5$ , CH-C(5)); 3.81 (dd,  $J = 2.5$ , 10.7, irrad. at  $4.19 \rightarrow$  change, CH' $-C(5)$ ); 3.83 (t,  $J = 7.5$ , irrad. at  $4.12 \rightarrow$  change, H $-C(6)$ ); 4.12  $(dd, J = 5.0, 7.2$ , irrad. at  $3.83 \rightarrow$  change, H $-C(7)$ ); 4.19 (ddd, J = 2.5, 5.4, 7.5, irrad. at  $3.83 \rightarrow$  change, H $-C(5)$ ); 4.39 (d, J = 11.7, PhCH); 4.44 (d, J = 11.7, PhCH); 4.46 (d, J = 11.2, PhCH); 4.64 (d, J = 11.2, PhCH); 4.78  $(d, J = 10.7, PhCH)$ ; 4.79  $(d, J = 11.3, 2 PhCH)$ ; 4.79  $(d, J = 5.1, \text{ irrad. at } 4.12 \rightarrow \text{change, H}-C(8))$ ; 5.06 (d,  $J = 11.7$ , PhCH); 7.06 – 7.44 (m, 30 arom. H); 7.76 (s, H – C(3)). <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 75 MHz): 58.31 (d, C(5)); 67.51 (t, CH<sub>2</sub>–C(5)); 72.11, 73.27, 73.61, 73.95 (4t, 4 PhCH<sub>2</sub>); 73.10, 75.96, 81.44 (3d, C(6), C(7), C(8));  $120.93 - 121.03$ , 124.91, 125.25 (several d of P(OPh)<sub>2</sub>); 128.49 (d, <sup>1</sup>J(C,P) = 256.4, C(2)); 127.71 – 129.49 (several d); 136.88, 137.29, 137.50, 137.90 (4s); 146.85  $(d, 3J(C,P) = 22.0, C(8a))$ ; 150.44  $(d, 2J(C,P) = 7.3)$ ; 150.49  $(d, {}^{2}J(C, P) = 7.3)$ . <sup>31</sup>P-NMR (CDCl<sub>3</sub>, 121 MHz): 5.24. FAB-MS: 793 (100,  $[M+1]$ <sup>+</sup>).

(5R,6R,7S,8S)-6,7,8-Tris(benzyloxy)-5-[(benzyloxy)methyl]-5,6,7,8-tetrahydro-5-methylimidazo[1,2-a]pyridinium Iodide (10). A soln. of 8 [19] (50 mg, 89.3  $\mu$ mol) in toluene (0.5 ml) was treated with MeI (30  $\mu$ ), 0.45 mmol) and kept at  $95^{\circ}$  for 4 h. Evaporation of the mixture followed by co-evaporation with toluene gave 64 mg of 10 (quant.). Coloured gel.  $R_f$  (AcOEt/iPrOH/H<sub>2</sub>O 10:3:0.5) 0.57. UV (CHCl<sub>3</sub>) 259 (3.6). IR (CHCl<sub>3</sub>): 3405w, 3155w, 3089w, 3067w, 3010w, 2946s, 2871m, 1602m, 1536m, 1497m, 1455s, 1364m, 1088s, 1001m, 909m. <sup>1</sup>H-NMR (CD<sub>3</sub>Cl, 300 MHz): 3.78 (dd, J = 2.9, 10.6, irrad. at  $4.50 \rightarrow d, J \approx 10.6$ , CH – C(5)); 3.97 (dd, J = 8.8, 10.6, irrad. at  $4.50 \rightarrow d, J = 10.3$ , CH'–C(5)); 4.10 (s, Me); 4.14 (dd, J = 4.1, 5.3, irrad. at  $4.50 \rightarrow d, J \approx 5$ ,  $H-C(6)$ ); 4.20 (dd, J = 3.5, 5.3, irrad. at 5.94  $\rightarrow$  d, J  $\approx$  5.6, H  $-C(7)$ ); 4.41 (d, J = 11.8, PhCH); 4.46 (d, J = 11.8,  $PhCH$ ); 4.48 – 4.52 (m, H – C(5)); 4.57 (d, J = 11.8, PhCH); 4.64 (d, J = 11.1, PhCH); 4.82 (d, J = 11.8, PhCH); 4.94  $(2d, J = 11.8, J = 11.1, 2 PhCH)$ ; 5.08  $(d, J = 11.1, PhCH)$ ; 5.94  $(d, J = 3.3, H - C(8))$ ; 7.08 - 7.41  $(m, 20 \text{ atom}, H)$ ; 7.44  $(d, J = 2.1, H - C(3))$ ; 7.49  $(d, J = 2.1, H - C(2))$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>, 75 MHz): 37.40  $(q, \text{Me})$ ; 61.94  $(d, \text{C}(5))$ ; 69.80  $(t, \text{CH}_2-\text{C}(5))$ ; 70.37, 72.75, 74.33 (3d, C(6), C(7), C(8)); 73.20, 73.56, 73.65, 73.70 (4t); 121.35 (d, C(3)); 124.42 (d, C(2)); 128.10 ± 128.78 (several d, arom. C ); 136.31, 136.36, 136.69, 136.88 (4s); 141.12 (s, C(8a)). FAB-MS: 575 (100,  $M^+$ ), 1277 (8,  $[2M+1]^+$ .

(5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,6-trihydroxy-5-(hydroxymethyl)-5-methylimidazo[1,2-a]pyridinium Iodide (11). A soln. of 9 (10 mg, 0.05 mmol) and MeI (4  $\mu$ l, 0.064 mmol) in DMSO (0.25 ml) was kept at 23° for 5 h. Evaporation gave 11 (17 mg, quant.). Colourless resin.  $R_f$  (AcOEt/MeOH 5:1) 0.04. <sup>1</sup>H-NMR  $(300 \text{ MHz}, (D_6)$ DMSO):  $3.69 - 3.79$  (m, irrad at  $4.79 \rightarrow \text{change}, \text{H}-\text{C}(6), \text{H}-\text{C}(7)$ ); 3.87 (s, Me); 3.82  $(dd, J=12.0, 4.2, CH-C(5))$ ; 4.04  $-4.08$   $(m, H-C(5))$ ; 4.79  $(d, J=6.0, H-C(8))$ ; 7.75, 7.77 (2d,  $J=3.0$ ,  $H-C(2)$ ,  $H-C(3)$ ). <sup>1</sup>H-NMR (300 MHz, D<sub>2</sub>O): 3.96 (s, Me); 3.99 – 4.02 (*m*, irrad. at 4.99  $\rightarrow$  change,  $H-C(6)$ ,  $H-C(7)$ ); 4.12 (dd, J = 12.8, 3.1, CH  $-C(5)$ ); 4.24 - 4.27 (m, H $-C(5)$ ); 4.30 (dd, J = 12.8, 2.5,  $CH-C(5)$ ; 4.98 (m, H $-C(8)$ ); 7.49, 7.65 (2d, J = 2.1, H $-C(2)$ , H $-C(3)$ ). <sup>13</sup>C-NMR (75 MHz, (D<sub>6</sub>)DMSO ) 35.56 (q, Me); 58.29 (t, CH<sub>2</sub>C(5)); 62.79 (d, C(5)); 66.19, 66.40 (2d, C(6), C(7)); 74.10 (d, C(8)); 119.51  $(d, C(2))$ ; 124.52  $(d, C(3))$ ; 143.97  $(s, C(8a))$ . FAB-MS: 215 (100,  $M^+$ ).

Diethyl (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2*phosphonate* (12). A mixture of  $5$  (250 mg; 0.36 mmol), AcOEt/MeOH/H<sub>2</sub>O 0.5:3:0.5 (1 ml), AcOH  $(0.5 \text{ ml})$ , and  $20\%$  Pd $(OH)$ <sub>2</sub> was hydrogenated at atmospheric pressure for 20 h. The suspension was filtered through *Celite*, washed with MeOH  $(20 \text{ ml})$  and chromatographed  $(ACOEt/PTOH/H<sub>2</sub>O 8:4:1)$ to give 12 as white powder after lyophilization (107 mg, 89%).  $R_f$  (AcOEt/PrOH/H<sub>2</sub>O 8:4:1) 0.3. IR (KBr): 3386s (br.), 3000s, 2985s, 1644m, 1528s, 1394s, 1221s, 1024s (br.), 860m, 797s, 697m, 662s, 603s. <sup>1</sup> H-NMR (D<sub>2</sub>O, 300 MHz): 1.30 (t, J = 7.2, 2 Me); 3.81 (dd, J = 9.2, 9.9, irrad. at  $4.63 \rightarrow d$ , irrad. at  $3.94 \rightarrow$  change,  $H - C(7)$ ); 3.94 (t, J = 9.2, H –  $C(6)$ ); 4.04 - 4.13 (m, irrad. at 3.94  $\rightarrow$  change, irrad. at 4.26  $\rightarrow$  change,  $H-C(5)$ ); CH $-C(5)$ ); 4.14 (q, J = 7.2, irrad. at 1.30  $\rightarrow$  change, 2 MeCH<sub>2</sub>); 4.26 (dd, J = 3.3, 13.8, CH' $-C(5)$ ); 4.63  $(d, J = 9.9, H - C(8))$ ; 7.93 (s, H – C(3)). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 15.49 (dq, <sup>3</sup>J(C,P) = 6.4, 2 MeCH<sub>2</sub>); 58.85 (t, CH<sub>2</sub>-C(5)); 64.19 (dt, <sup>2</sup>J(C,P) = 4.8, 2 MeCH<sub>2</sub>); 60.97 (d, C(5)); 67.23, 67.96, 74.53 (3d, C(6), C(7),  $C(8)$ ); 127.52 (d, <sup>1</sup>J(C,P) = 247.4, C(2)); 128.47 (dd, <sup>2</sup>J(C,P) = 36.7, C(3)); 150.32 (d, <sup>3</sup>)  $^{31}P\text{-NMR}$  (D<sub>2</sub>O, 121 MHz): 14.62. FAB-MS: 337 (100,  $[M+1]^+$ ), 359 (61,  $[M+Na]^+$ ), 695 (13,  $[2M + Na]^{+}$ ).

Diethyl (5R,6R,7S,8S)-6,7,8-Tris{[(tert-Butyl)dimethylsilyl]oxy}-5-({[(tert-Butyl)dimethylsilyl]oxy}meth $y$ l)-5,6,7.8-tetrahydroimidazo[1,2-a]pyridine-2-phosphonate (13). A soln. of 12 (30 mg, 89.3 µmol) in DMF (0.15 ml) was treated with <sup>t</sup> BuMe2SiCl (188 mg, 1.25 mmol) and 1H-imidazole (170 mg, 2.5 mmol) and stirred at 25 $\degree$  for 48 h. The mixture was diluted with Et<sub>2</sub>O, washed by H<sub>2</sub>O. Combined org. phase was dried (MgSO<sub>4</sub>), filtered, and evaporated. FC (hexane/AcOEt 7:3) gave 13 (61 mg, 87%). White solid.  $R_f$  (hexane/AcOEt 3:7) 0.75. IR (CCl<sub>4</sub>): 2955s, 2930s, 2897m, 2858s, 1515w, 1472m, 1463w, 1442w, 1390w, 1362w, 1259s, 1236m, 1188w,  $1099s, 1034m, 1006w, 970w, 940m, 838s, 668w, 594w.$   $\rm{H\text{-}NMR}$  ( $\rm{C_6D_6}$ , 500 MHz): 0.01, 0.03, 0.08, 0.10, 0.13, 0.15, 0.15, 0.25, 0.40 (8s, 8 Me); 0.85, 0.92, 0.93; 1.03 (4s, Me<sub>3</sub>CSi); 1.16 (dt,  $J = 0.5, 7.0$ , Me); 1.18 (dt,  $J = 0.5, 7.0, 2$  Me); 3.82 (dd, J = 7.0, 11.0, CH – C(5)); 3.87 (dd, J = 4.5, 11.0, CH'–C(5)); 4.10 – 4.28 (m, 2 MeCH<sub>2</sub>, H – C(5), H – C(6); 4.29 (t, J = 2.2, irrad. at 5.0  $\rightarrow$  d, H – C(7)); 5.00 (dd, J = 0.7, 2.2, H – C(8)); 7.87 (s, H – C(  $^{13}$ C-NMR (C<sub>6</sub>D<sub>6</sub>, 125 MHz):  $-5.53$ ,  $-5.33$ ,  $-4.88$ ,  $-4.84$ ,  $-4.55$ ,  $-4.51$ ,  $-4.22$ ,  $-4.03$ ,  $-3.99$  (8t, 8 Me); 16.53  $(dq, {}^{3}J(C, P) = 5.2, Me); 16.56 (dq, {}^{3}J(C, P) = 5.0, Me); 18.1, 18.19, 18.35, 18.54 (4s, 4 Me<sub>3</sub>CSi); 25.85, 25.95, 26.01,$ 26.29 (4q, 4 Me<sub>3</sub>CSi); 61.83 (t, <sup>2</sup>J(C,P) = 5.1, MeCH<sub>2</sub>); 61.93 (t, <sup>2</sup>J(C,P) = 5.3, MeCH<sub>2</sub>); 63.43 (t, CH<sub>2</sub>-C(5)); 63.73  $(d, C(5))$ ; 69.64, 71.63, 77.54  $(3d, C(6), C(7), C(8))$ ; 129.12  $(dd, \frac{2J(C, P)}{36.9}, C(3))$ ; 130.99  $(d, {}^{1}J(C, P) = 241.2, C(2))$ ; 146.92  $(d, {}^{3}J(C, P) = 22.0, C(8a))$ .  ${}^{31}P\text{-NMR } (C_6D_6, 202 \text{ MHz})$ : 11.56. FAB-MS: 735  $(84, [M+1-\text{Bu}]^+)$ , 793 (100,  $[M+1]^+$ ). Anal. calc. for  $C_{36}H_{77}N_2O_7Si_4P$  (793.34): C 54.50, H 9.78, N 3.53, P 3.90; found: C 54.54, H 9.66, N 3.54, P 3.84.

Ethyl Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate (14). A soln. of 12 (14 mg, 41.6 µmol) in 1M aq. NaOH (200 µl) was stirred at 70 $^{\circ}$  for 14 h, neutralized with Amberlite IRC 50 ( $H^+$  form), filtered, and evaporated. A soln. of the residue in MeOH (5 ml) was adsorbed on silica gel. FC (PrOH/H<sub>2</sub>O 8:2) gave 14 as white powder after lyophilization (7.5 mg, 58%).  $R_f$  $(iPrOH/H<sub>2</sub>O 7:3) 0.5.$ <sup>1</sup>H-NMR (D<sub>2</sub>O, 300 MHz): 1.96 (t, J = 7.3, irrad. at 3.85  $\rightarrow$  s, Me); 3.80 (t, J = 9.6, irrad. at  $4.26 \rightarrow$  change, irrad. at  $3.94 \rightarrow$  change, H $-C(7)$ );  $3.85$  (q,  $J = 7.3$ , MeCH<sub>2</sub>);  $3.94$  (t,  $J = 9.6$ , H $-C(6)$ );  $4.00 - 4.08$  $(m, \text{irrad. at } 4.26 \rightarrow \text{change, irrad. at } 3.94 \rightarrow \text{change, H-C(5)}); 4.07 (dd, J = 2.8, 13.0, \text{irrad. at } 4.26 \rightarrow \text{change, h}$  $CH-C(5)$ ; 4.26 (dd, J = 2.3, 13.0, CH'  $-C(5)$ ); 4.62 (d, J = 9.6, H $-C(8)$ ); 7.57 (s, H $-C(3)$ ). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 15.77  $(dq, {}^{3}J(C,P) = 6.4,$  Me); 58.75  $(t, CH_2-(5))$ ; 64.37  $(dt, {}^{2}J(C,P) = 4.8,$  MeCH<sub>2</sub>); 60.56  $(d, C(5))$ ; 67.32, 68.11, 74.78 (3d, C(6), C(7), C(8)); 124.55 (dd, <sup>2</sup>J(C,P) = 31.9, C(3)); 134.77 (d, <sup>1</sup>J(C,P) = 226.7, C(2)); 148.64  $(d, {}^{3}J(C, P) = 19.1, C(8a))$ . <sup>31</sup>P-NMR (D<sub>2</sub>O, 121 MHz): 8.30. FAB-MS (glycerol): 309 (9.40,  $[M + 1]$ <sup>+</sup>), 331 (5.72,  $[M + Na]$ <sup>+</sup>).

Ethyl Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate Hydrochloride (14  $\cdot$  HCl). A soln. of 14 (10 mg, 32 µmol) in D<sub>2</sub>O (0.7 ml) was treated with 1m aq. HCl (20  $\mu$ ), evaporated, co-evaporated with H<sub>2</sub>O, and lyophilized. The residue was taken up in H<sub>2</sub>O (1 ml), treated with  $Bio-Rad AG 2-X8$  resin (Cl<sup>-</sup> form), filtered, and lyophilized to give 8 mg of 14 <sup>·</sup> HCl. <sup>1</sup>H-NMR (D<sub>2</sub>O, 300 MHz): 1.23 (t, J = 7.0, MeCH<sub>2</sub>); 3.92 (q, J = 7.2, MeCH<sub>2</sub>); 3.93 (br. t, J = 9.6, irrad. at  $4.89 \rightarrow d, J \approx 10, H-C(7)$ ;  $4.03$  (t, J = 9.2, irrad. at  $4.26 \rightarrow d, J \approx 10, H-C(6)$ );  $4.11$  (dd, J = 3.2, 12.8, irrad. at  $4.26 \rightarrow d, J \approx 10$ , CH-C(5));  $4.22 - 4.29$  (m, H-C(5)); 4.32 (dd, J = 2.4, 12.8, irrad. at  $4.26 \rightarrow$  change, CH' – C(5)); 4.89  $(d, J = 8.4, H - C(8))$ ; 7.84 (s, H – C(3)). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 18.45  $(dq, {}^{3}J(C, P) = 6.1$ , Me); 61.15 (t, CH<sub>2</sub>-C(5)); 65.19 (dt, <sup>2</sup>J(C,P)=6.1, MeCH<sub>2</sub>); 65.28 (d, C(5)); 69.31, 69.68, 75.99 (3d, C(6),  $C(7)$ ,  $C(8)$ ); 127.89  $(dd, {}^{2}J(C,P) = 20.7$ ,  $C(3)$ ); 131.59  $(d, {}^{1}J(C,P) = 202.6$ ,  $C(2)$ ); 150.99  $(d, {}^{3}J(C,P) = 7.3$ ,  $C(8a)$ ). <sup>31</sup>P-NMR (D<sub>2</sub>O, 121 MHz):  $-1.78$ .

Phenyl Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate (15). A soln. of  $7(17 \text{ mg}, 21.4 \text{ \mu}$ mol) in MeOH/H<sub>2</sub>O/AcOH  $2:0.5:1(1.75 \text{ ml})$  was treated with 20% Pd(OH) $_2$ /C (17 mg) and hydrogenated at atmospheric pressure for 34 h. The suspension was filtered through Celite, and the residue washed with MeOH/H<sub>2</sub>O 9:1. Evaporation of the filtrate gave 10 mg of crude which was taken up in H<sub>2</sub>O (1 ml) treated with *Dowex 50W8* (H<sup>+</sup> form), filtered, and lyophilized. The residue was taken up in MeOH/H<sub>2</sub>O 1:1 (1.5 ml), treated with activated charcoal, filtered, and lyophilized to give 10 mg of white solid **15**.  $R_f$  (<sup>i</sup>PrOH/H<sub>2</sub>O 8 :2) 0.32. <sup>1</sup>H-NMR (D<sub>2</sub>O, 300 MHz): 3.81 (t, J = 8.8, irrad. at 4.65  $\rightarrow$  d, J  $\approx$  9.3,  $H-C(7)$ ); 3.94 (t, J = 8.8, H  $-C(6)$ ); 3.98 - 4.11 (m, H  $-C(5)$ , CH  $-C(5)$ ); 4.21 (br. d, J = 12.8, CH'  $-C(5)$ ); 4.65  $(d, J = 8.8, H - C(8))$ ; 7.03 (br.  $d, J = 7.8, 2$  arom. H); 7.16 (br. t,  $J = 7.8,$  arom. CH); 7.32 (br. t,  $J = 7.8,$ 2 arom. H); 7.52 (s, H–C(3)). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 61.53 (t, CH<sub>2</sub>–C(5)); 63.32 (d, C(5)); 70.12, 70.89, 77.52 (3d, C(6), C(7), C(8)); 124.18, 124.23, 127.34 (3d, arom. CH); 128.03 (dd, <sup>2</sup>J(C,P) = 34.2, C(3)); 132.58 (d, arom. CH); 137.18 (d, <sup>1</sup>J(C,P) = 233.17, C(2)); 151.66 (d, <sup>3</sup>J(C,P) = 20.7, C(8a)); 154.5 (d, <sup>2</sup>J(C,P) = 7.3).<br><sup>31</sup>P-NMR (D<sub>2</sub>O, 121 MHz): 6.05, ESI-MS (negative mode): 355 (100, [M – 1]<sup>-</sup>).

Triethylammonium Hydrogen (5R,6R,7S,8S)-6,7,8-Trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate (17). A soln. of 5 (260 mg, 0.373 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3.5 ml) at 0° was treated with Me<sub>3</sub>SiBr  $(0.29 \text{ ml}, 2.24 \text{ mmol})$ , warmed to  $25^{\circ}$ , and stirred for 16 h. The mixture was concentrated and co-evaporated with toluene  $(4 \times 5 \text{ ml})$ . The residue was taken up in MeOH/H<sub>2</sub>O 9 : 1 (2 ml), evaporated, and co-evaporated with toluene until crude 16 became a foam (241 mg). A soln. of crude 16 in MeOH/AcOEt/H<sub>2</sub>O 3:1:1 (4 ml) was treated with 20% Pd(OH) $_2$ /C (180 mg) and hydrogenated at atmospheric pressure for 19 h. The suspension was filtered through *Celite*, and the residue washed with MeOH/H<sub>2</sub>O 95 : 5 (75 ml). Evaporation of the filtrate gave 116.5 mg of crude 17, which was taken up in 1 ml of H<sub>2</sub>O and applied to a DEAE-cellulose column (Cellex-D, *Bio-rad*,  $18 \times 1.5$  cm; UV detection). The column was washed with H<sub>2</sub>O (30 ml), and 17 was eluted with a triethylammonium hydrogen carbonate buffer ( $pH \approx 7$ ; 5 mm, 30 ml; 10 mm, 40 ml; 20 mm, 40 ml). The fractions containing **17** were combined and lyophilized  $(3 \times)$  to give **17** (110 mg, 81%, 0.83 equiv. of Et<sub>3</sub>NH<sup>+</sup>).

Triethylammonium salt 17 was also obtained after treatment of an aq. soln. of crude deproteced free acid with  $Et<sub>3</sub>N$  (3 equiv.). The soln. was evaporated, concentrated, and co-evaporated with toluene. The residue was taken up in H<sub>2</sub>O/MeOH ca. 1:1, treated with activated charcoal, and filtered. Evaporation and lyophilization  $(3 \times)$  gave 17 (0.87 equiv. of  $Et_3NH^+$ ).

Data of (5R,6R,7S,8S)-6,7,8-Tris(benzyloxy)-5-[(benzyloxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2-a]pyridine-2-phosphonic Acid (16):  $R_f$  (RP C18, MeOH/H<sub>2</sub>O 9:1) 0.6. IR (CHCl<sub>3</sub>): 3500 – 2233w (br.), 3028m, 2971m, 1602w, 1496m, 1454m, 1363m, 1230m, 1094s, 996s. <sup>1</sup>H-NMR (CD<sub>3</sub>OD, 300 MHz): 3.78 (dd, J = 6.5, 10.6, CH $-C(5)$ ; 3.90 (dd, J = 3.2, 10.6, CH' $-C(5)$ ); 4.20 (dd, J = 5.7, 9.7, H $-C(6)$ ); 4.22 (br. t, J = 4.9, irrad. at  $5.01 \rightarrow d, J \approx 5.4$ , H $-C(7)$ ); 4.42 (d, J = 11.8, PhCH); 4.50 (d, J = 11.8, PhCH); 4.52 (d, J = 11.4, PhCH); 4.60  $\rightarrow$ 4.72 (m, irrad. at  $3.90 \rightarrow$  change, H $-C(5)$ , 3 PhCH); 4.80 (d, J = 11.4, PhCH); 4.91 (d, J = 11.4, PhCH); 5.01  $(d, J = 4.1, \text{ irrad. at } 4.22 \rightarrow s, \text{ H} - \text{C}(8))$ ; 7.12 - 7.39 (m, 20 arom. H); 7.70 (d, J = 2.5, H - C(3)). <sup>13</sup>C-NMR  $(CD_3OD, 75 MHz)$ : 62.18  $(d, C(5))$ ; 69.06  $(t, CH_2-C(5))$ ; 72.86, 74.77, 78.43 (3d, C(6), C(7), C(8)); 74.20, 74.28, 74.64, 75.03 (4t, 4 PhCH<sub>2</sub>); 127.86 (dd, <sup>2</sup>J(C,P) = 22.0, C(3)); 130.03 (d, <sup>1</sup>J(C,P) = 214.0, C(2)); 129.21 –  $129.73$  (several d));  $138.43 - 138.71$  (4s);  $146.52$  (d,  $3J(C,P) = 8.5$ , C(8a)).  $31P-NMR$  (CD<sub>3</sub>OD, 121 MHz):  $-1.63$ . FAB-MS: 663 (32,  $[M + Na]$ <sup>+</sup>); 685 (100,  $[M - 1 + 2 Na]$ <sup>+</sup>); 707 (43,  $[M - 2 + 3 Na]$ <sup>+</sup>; 1347 (16, [2  $(M - 1) + 3$  $\text{Na}^{\dagger}$ ).

Data of 17:  $R_f$  (MeOH/NH<sub>3</sub>/H<sub>2</sub>O 4:3:1) 0.57. UV (H<sub>2</sub>O): 232 (2.78). IR (KBr): 3386s (br.), 2361m, 1654w,  $1476m$ ,  $1398m$ ,  $1109s$  (br.),  $903m$ ,  $667m$ ,  $592s$ ,  $492m$ .  ${}^{1}$ H-NMR (D<sub>2</sub>O, 300 MHz):  $1.27$  (t,  $J = 7.6$ , 3  $MeCH_2$ );  $3.19$  $(q, J = 7.6, 3 \text{ MeCH}_2)$ ; 3.90  $(t, J = 9.7, \text{irrad. at } 4.83 \rightarrow \text{change}, \text{H}-\text{C}(7))$ ; 4.02  $(dd, J = 8.8, 9.7, \text{irrad. at } 3.90 \rightarrow$ change, H $-C(6)$ ); 4.09 (dd, J = 3.1, 13.2, CH $-C(5)$ ); 4.19 (br. d, J = 8.8, H $-C(5)$ ); 4.30 (dd, J = 2.2, 13.2, CH'–C(5); 4.83 (d, J=9.7, irrad. at 3.90  $\rightarrow$  change, H–C(8)); 7.60 (s, H–C(3)). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 10.99 (q, 3 Me); 49.46 (t, 3 MeCH<sub>2</sub>); 61.01 (t, CH<sub>2</sub>-C(5)); 64.66 (d, C(5)); 69.52, 69.69, 76.29 (3d, C(6), C(7),  $C(8)$ ); 125.02 (dd, <sup>2</sup>J(C,P) = 22.0, C(3)); 137.40 (d, <sup>1</sup>J(C,P) = 190.4, C(2)); 149.59 (d, <sup>3</sup>J(C,P) = 8.8, C(8a)). <sup>31</sup>P-NMR (D<sub>2</sub>O, 121 MHz):  $-1.97$ . ESI-MS (MeOH/H<sub>2</sub>O 1 : 1, 1% AcOH, negative mode): 279 ( $[M-1]$ <sup>-</sup>), 339  $([M + AcO]^{-}),$  559  $([2M - 1]^{-}),$  839  $([3 M - 1]^{-}).$ 

Disodium (5R,6R,7S,8S)-6,7,8-Trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate (18). A soln. of 17 (60 mg, 0.164 mmol, 0.75 equiv. of  $Et_3NH^+$ ) in H<sub>2</sub>O (5 ml) was treated with *Dowex 50W8* (Na<sup>+</sup> form) then with Dowex CCR-2 (Na<sup>+</sup> form), filtered, and lyophilized to give **18** (38 mg) as a white solid, of which 25 mg were taken up in a minimum of  $H<sub>2</sub>O$  (ca. 0.2 ml) and treated with MeOH (1.5 ml) until a white precipitate was formed. The mixture was kept for 12 h at  $4^\circ$ . The precipitate was isolated by centrifugation to give, after washing with MeOH and drying under vacuum, 20 mg of 18. White powder. UV  $(H_2O)$ : 230 (3.10). IR (KBr): 3384s (br.), 1648m, 1523m, 1438m, 1334m, 1069s (br.), 996s, 952s, 906m, 667s, 602s, 498s. <sup>1</sup> H-NMR (D2O,

500 MHz): 3.76  $(dd, J = 9.0, 10.0, \text{irrad. at } 4.61 \rightarrow \text{change, } H - C(7))$ ; 3.90  $(t, J = 10.0, H - C(6))$ ; 3.96 - 4.01  $(m, H-C(5))$ ; 4.01 (dd, J = 2.5, 13.0, CH – C(5)); 4.21 (dd, J = 2.5, 13.0, CH'–C(5)); 4.61 (d, J = 9.0, H–C(8)); 7.31 (s, H $-C(3)$ ). <sup>13</sup>C-NMR (D<sub>2</sub>O, 125 MHz): 61.08 (t, CH<sub>2</sub> $-C(5)$ ); 63.26 (d, C(5)); 69.82, 70.38, 77.13 (3d,  $C(6)$ ,  $C(7)$ ,  $C(8)$ ); 123.69 (dd, <sup>2</sup>J(C,P) = 26.8, C(3)); 142.41 (d, <sup>1</sup>J(C,P) = 202.6, C(2)); 149.29 (d, <sup>3</sup>J(C,P) = 14.6, C(8a)). <sup>31</sup>P-NMR (D<sub>2</sub>O, 202 MHz): 2.64. FAB-MS (glycerine, negative mode): 279 (100,  $[M - 2\text{ Na}]$ <sup>-</sup>), 301 (59,  $[M - Na]$ <sup>-</sup>). Anal. calc. for C<sub>8</sub>H<sub>11</sub>N<sub>2</sub>Na<sub>2</sub>O<sub>7</sub>P  $\cdot$  0.75 H<sub>2</sub>O (337.64): C 28.46, H 3.73, N 8.29; found: C 28.64, H 3.67, N 8.14.

Triethylammonium Hydrogen (5R,6R,7S,8S)-6,7,8-Triacetoxy-5-[(acetoxy)methyl]-5,6,7,8-tetrahydroimida $zol(1,2-a)$  pyridine-2-phophonate (19). A soln. of 17 (0.70 equiv. of Et<sub>3</sub>NH<sup>+</sup>; 15 mg, 40 µmol) in pyridine (0.5 ml) was treated with Ac<sub>2</sub>O (195  $\mu$ ) and stirred at 25° overnight. The soln. was evaporated and co-evaporated with toluene. The phosphonate 19 (23 mg,  $0.72$  equiv. of  $Et\rightarrow NH^+$ ) was used for the next step without further purification. R<sub>f</sub> (AcOEt/MeOH/H<sub>2</sub>O 8:4:1) 0.5. UV (H<sub>2</sub>O): 289 (2.28), 234 (2.89). IR (KBr): 3444m, 2677w, 2360w, 1748s, 1652w, 1435w, 1373m, 1229s, 1036s, 920w, 838w. <sup>1</sup>H-NMR (CD<sub>3</sub>OD, 300 MHz): 1.29 (t, J = 7.9,  $3 \text{ MeCH}_2$ ); 2.06, 2.08, 2.09, 2.11 (4s, 4 AcO); 3.19 (q, J = 7.9, 3 MeCH<sub>2</sub>); 4.42 (dd, J = 5.4, 12.6, CH – C(5)); 4.59  $(dd, J=3.8, 12.6, CH'-C(5))$ ; 4.67 – 4.76  $(m, H-C(5))$ ; 5.54 (br. t,  $J \approx 6.5$ , irrad. at  $4.7 \rightarrow d, J \approx 9.0, H-C(6))$ ; 5.56 (br. t,  $J \approx 6.0$ , irrad. at  $6.16 \rightarrow d$ ,  $J \approx 7.5$ , H $-C(7)$ ); 6.16 (dd, J = 5.1, 0.7, H $-C(8)$ ); 7.74 (d, J = 1.2,  $H-C(3)$ ). <sup>13</sup>C-NMR (D<sub>2</sub>O, 75 MHz): 8.25 (q, 3 MeCH<sub>2</sub>); 20.29, 20.28, 21.96, 22.06 (4q, 4 Me); 46.75  $(t, 3 \text{ MeCH}_2)$ ; 56.53  $(d, C(5))$ ; 62.15  $(t, CH_2-C(5))$ ; 66.11, 66.69, 70.43 (3d, C(6), C(7), C(8)); 126.60  $(dd, {}^{2}J(C, P) = 36.7, C(3))$ ; 135.12  $(d, {}^{1}J(C, P) = 236.2, C(2))$ ; 142.62  $(d, {}^{3}J(C, P) = 20.6, C(8a))$ ; 172.51 – 173.66 (several s,  $4 C=O$ ). <sup>31</sup>P-NMR (CD<sub>3</sub>OD, 121 MHz):  $-1.12$ . FAB-MS: 102 (72, Et<sub>3</sub>NH<sup>+</sup>), 449 (29, [M + 1]<sup>+</sup>), 550  $(15, [(M + Et<sub>3</sub>NH]<sup>+</sup>), 592 (100, [M - 3 + Et<sub>3</sub>N + 2 Na]<sup>+</sup>), 693 (12, [M - 3 + 2 Et<sub>3</sub>N + 2 Na]<sup>+</sup>).$ 

Oleyl6) Hydrogen (5R,6R,7S,8S)-6,7,8-Triacetoxy-5-[(acetoxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2 a]pyridine-2-phosphonate (20). A soln. of 19 (39 mg, 0.068 mmol) and oleyl alcohol (16 mg, 0.059 mmol) in pyridine (1 ml) was treated at  $25^{\circ}$  with CCl<sub>3</sub>CN (0.1 ml, 1 mmol) and warmed to 70°. After 13 h, the resulting brown soln. was evaporated and co-evaporated with toluene, dissolved in AcOEt  $(10 \text{ ml})$ , and washed  $(H<sub>2</sub>O)$ . The aq. phase was extracted with AcOEt  $(4 \times 10 \text{ ml})$ . The combined org. phases were dried  $(MgSO_4)$ , filtered, and evaporated. FC (silica gel 60; AcOEt/PrOH/H<sub>2</sub>O  $10:1:0.1 \rightarrow 10:4:1$ ) followed by *RP 18* (MeOH/H<sub>2</sub>O  $8:2 \rightarrow 95:5$ ) gave 20 (27 mg) as a brown gel. A soln. of 20 in MeOH (2 ml) was treated with activated charcoal and gave, after filtration and evaporation, 20 (20 mg, 50% from oleyl alcohol). Solid.  $R_f$  (AcOEt/MeOH/H<sub>2</sub>O 8: 4: 1) 0.6. UV (CHCl<sub>3</sub>): 273 (2.24), 245 (2.64). IR (CCl<sub>4</sub>): 3341w, 3135w, 2927s, 2855m, 1763s, 1512w, 1466w,  $1433w$ ,  $1369w$ ,  $1260m$ ,  $1222s$ ,  $1065s$ ,  $946w$ ,  $904w$ .  $1H\text{-NMR (CD<sub>3</sub>OD, 300 MHz): 0.89(t, J = 6.4, Me); 1.29 - 1.40$  $(m, 22 H)$ ; 1.46 - 1.58  $(m, CH, CH, O)$ ; 1.92 - 2.25  $(m, CH, CH, CH)$ ; 2.06, 2.07, 2.09, 2.10 (4s, 4 AcO); 3.76  $(q, J = 6.4, \text{irrad. at } 1.53 \rightarrow d, \text{CH}_2CH_2O$  ); 4.43 (dd,  $J = 5.4, 12.7, \text{CH}-C(5)$ ); 4.57 (dd,  $J = 3.6, 12.7, \text{CH}'-C(5)$ ); 4.54 (br.  $q, J \approx 4.5$ , H-C(5)); 5.28 - 5.42 (m, irrad. at 2.20  $\rightarrow$  change, CH<sub>2</sub>CH=CHCH<sub>2</sub>); 5.51 (t, J = 7.0,  $H-C(6)$ ); 5.55 (dd, J = 5.5, 7.5, irrad. at 6.09  $\rightarrow$  d, J  $\approx$  8.0, H $-C(7)$ ); 6.09 (d, J = 5.5, H $-C(8)$ ); 7.59 (d, J = 0.9,  $H-C(3)$ ). <sup>13</sup>C-NMR (CD<sub>3</sub>OD, 75 MHz): 14.38 (q, Me); 20.49 – 20.82 (several q); 23.68 – 30.86 (several t); 31.87  $(t, {}^{3}J(C, P) = 7.3, CH_2CH_2O)$ ; 30.03  $(t, CH_2)$  58.14  $(d, C(5))$ ; 63.17  $(t, CH_2-C(5))$ ; 67.52, 67.76, 71.76 (C(6),  $C(7)$ ,  $C(8)$ ); 65.88  $(dt, {}^{2}J(C, P) = 6.1$ ,  $CH_{2}CH_{2}O$ ); 125.99  $(dd, {}^{2}J(C, P) = 30.5$ ,  $C(3)$ ); 131.00  $(d, CH=CH)$ ; 139.00 (d, <sup>1</sup>J(C,P) = 225.8, C(2)); 143.47 (d, <sup>3</sup>J(C,P) = 18.0, C(8a)); 171.01, 171.22, 171.61, 172.00 (4s, 4 C=O).<br><sup>31</sup>P-NMR (CD<sub>3</sub>OD): 3.98. FAB-MS: 721 (100, [M + Na]<sup>+</sup>).

Phytanyl6) Hydrogen (5R,6R,7S,8S)-6,7,8-Triacetoxy-5-[(acetoxy)methyl]-5,6,7,8-tetrahydroimidazo[1,2 a/pyridine-2-phosphonate  $(21)$ . A soln. of 19 (22 mg, 42.7 µmol) in pyridine  $(0.5 \text{ ml})$  was treated with phytanol  $(10.2 \text{ mg}, 34 \text{ µmol})$  and CCl<sub>3</sub>CN  $(8.6 \text{ µ}, 85.4 \text{ µmol})$ , and stirred at  $60^{\circ}$  for 18 h. After workup as described for 20, FC (CHCl<sub>3</sub>  $\rightarrow$  CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 10 : 1.2 : 0.1  $\rightarrow$  10 : 1.5 : 0.1  $\rightarrow$  10 : 2 : 0.2) gave 21 containing *ca*. 15% of the deacetylated product at  $C(8)$  (determined by <sup>1</sup>H-NMR). Acetylation of the mixture (1.2 ml of pyridine/Ac<sub>2</sub>O 5 : 1, 25°, overnight) gave 21 (15 mg, 61% from phytanol). Solid.  $R_f$  (CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 10: 2: 0.25) 0.24. IR (CCl4): 3353w, 3138w, 2956s, 2927s, 2868m, 1763s, 1512w, 1463m, 1432w, 1369m, 1222s, 1065s, 946s, 903w. UV (CDCl<sub>3</sub>): 274 (1.96), 243 (2.49). <sup>1</sup>H-NMR (CD<sub>3</sub>OD, 300 MHz): 0.80 (*d*, *J* = 6.4, irrad. at 1.54  $\rightarrow$  change, Me); 0.82  $(d, J = 6.4, \text{ irrad. at } 1.54 \rightarrow \text{change, Me})$ ; 0.86, 0.87, 0.88  $(3d, J = 6.4, 3 \text{ Me})$ ; 1.00 - 1.45  $(m, 21 \text{ H})$ ; 1.53  $(sept.$ ,  $J = 6.5$ , 1 H ); 1.47 - 1.66  $(m, 2, H)$ ; 2.06, 2.07, 2.09, 2.10  $(4s, 4 \text{ AcO})$ ; 3.72 - 3.90  $(m, CH_2OP)$ ; 4.43  $(dd, J = 5.1, 12.1, \text{ irrad. at } 4.90 \rightarrow d, J \approx 12.5, \text{ CH} - \text{C}(5)$ ; 4.58  $(dd, J = 3.3, 12.1, \text{ irrad. at } 4.90 \rightarrow d, J \approx 12.5,$ 

<sup>6)</sup> Oleyl =  $(Z)$ -octadec-9-enyl; phytanyl = 3,7,11,15-tetramethylhexadecyl; dolichyl-19 = 3,7,11,15,19,23,27,31, 35,39,43,47,51,55,59,63,67,71,75-nonadecamethylhexaheptaconta-6,10,14,18,22,26,30,34,38,42,46,50,54,58,62, 66,70,74-octadecaenyl.

CH'  $-C(5)$ ); 4.90 (br. q,  $J \approx 4.6$ , H $-C(5)$ ); 5.51 (t, J = 6.5, irrad. at 4.90  $\rightarrow$  d, J  $\approx 8.0$ , H $-C(6)$ ); 5.56 (dd, J = 4.7, 6.5, irrad. at  $6.09 \rightarrow d, J \approx 8$ , H $-C(7)$ ); 6.09 (dd, J = 0.9, 4.7, H $-C(8)$ ); 7.60 (s, H $-C(3)$ ). <sup>13</sup>C-NMR (CD<sub>3</sub>OD, 75 MHz):  $19.69 - 23.07$  (several q, 9 Me);  $25.47$ ,  $25.85$  (2t);  $29.10 - 33.95$  (several d);  $38.37 - 40.52$  (several t); 63.17 (t, CH<sub>2</sub>–C(5)); 64.19 (m, CH<sub>2</sub>OP); 58.20 (d, C(5)); 67.46, 67.65, 71.57 (3d, C(6), C(7), C(8)); 126.09  $(dd, {}^2J(C,P) = 30.5, C(3))$ ; 138.68  $(d, {}^1J(C,P) = 225.8, C(2))$ ; 143.48  $(d, {}^3J(C,P) = 19.5, C(8a))$ ; 170.93, 171.16, 171.58, 171.92 (4s, 4 C=O). <sup>31</sup>P-NMR (CD<sub>3</sub>OD, 121 MHz): 3.78. FAB-MS: 751 (100,  $[M + Na]$ <sup>+</sup>). Anal. calc. for  $C_{36}H_{61}N_2O_{11}P \cdot 2 H_2O$  (764.12): C 56.53, H 8.56, N 3.66; found: C 56.36, H 8.44, N 3.65.

Dolichyl-19<sup>6</sup>) Hydrogen (5R,6R,7S,8S)-6,7,8-Triacetoxy-5-I (acetoxy)methyll-5,6,7,8-tetrahydroimida $z_0$ [1,2-a]pyridine-2-phosphonate (22). A soln. of 19 (22 mg, 42.7 µmol) in pyridine (0.4 ml) was treated with dolichol-19 (10 mg, 7.6 µmol) and CCl<sub>3</sub>CN (9 µl, 90 µmol) and stirred at  $60^{\circ}$  for 18 h. The mixture was evaporated and co-evaporated with toluene, diluted with CHCl<sub>3</sub>, washed with H<sub>2</sub>O, dried (MgSO<sub>4</sub>), and filtered. Evaporation and FC (CHCl<sub>3</sub>  $\rightarrow$  CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 10:0.6:0.05  $\rightarrow$  10:0.8:0.05  $\rightarrow$  10:1.0:0.05) followed by acetylation (0.6 ml pyridine/Ac<sub>2</sub>O 5 : 1, 25°, overnight) gave 22 (10 mg, 65% from dolichol-19). Coloured gel.  $R_f$ (CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 10:2:0.25) 0.31. IR (CCl<sub>4</sub>): 3317w, 2962s, 2928s, 2955m, 1762s, 1664w, 1449m, 1376m, 1261s, 1222s, 1088s, 1031s. <sup>1</sup>H-NMR (CD<sub>3</sub>OD/CDCl<sub>3</sub> 4 : 2, 500 MHz): 0.84 (d, J = 6.5, Me); 1.10 – 1.45 (m, 15 H instead of the expected 5 H); 1.62, 1.60 (2s, 4 Me); 1.68 (s, 15 Me); 1.93 – 2.08 (m, 70 H); 2.09, 2.10, 2.11, 2.12 (4s,  $4 \text{ AcO}$ ); 3.66 – 3.84 (m, CH<sub>2</sub>OP); 4.43 (dd, J = 5.5, 12.5, irrad. at  $4.63 \rightarrow d$ , J  $\approx$  12.8, CH – C(5)); 4.56 (dd, J = 3.5, 12.5, irrad. at  $4.63 \rightarrow d$ ,  $J \approx 12.8$ , CH' $-C(5)$ );  $4.63$  (br. q,  $J \approx 5.0$ , H $-C(5)$ ); 5.46 (dd,  $J = 6.5$ , 7.5, irrad. at  $4.63 \rightarrow$ d, J  $\approx$  7.8, H – C(6)); 5.54 (dd, J = 5.7, 7.5, irrad. at 6.07  $\rightarrow$  d, J  $\approx$  7.5, H – C(7)); 6.07 (d, J = 5.5, H – C(8)); 7.57  $(s, H-C(3))$ . <sup>13</sup>C-NMR (CD<sub>3</sub>OD/CDCl<sub>3</sub> 4 : 2, 125 MHz): 16.26 – 20.86 (several q); 23.74, 23.85 (2q); 25.93 – 27.44 (several t); 29.89 (d); 32.59 – 40.43 (several t); 57.56 (d, C(5)); 62.64 (t, CH<sub>2</sub>–C(5)); 63.71 (t, CH<sub>2</sub>OP); 66.85, 67.12, 71.12 (3d,  $C(6)$ ,  $C(7)$ ,  $C(8)$ ); 124.97 - 126.55 (several d); 131.65 - 135.87 (several s); 138.29  $(d, {}^{1}J(C, P) = 224.0, C(2))$ ; 142.50  $(d, {}^{3}J(C, P) = 18.7, C(8a))$ ; 170.23, 170.42, 170.79, 171.29  $(4s, 4C=O)$ . <sup>31</sup>P-NMR  $(CD_3OD/CDCI_3 4:2, 202 MHz)$ : 3.87. FAB-MS: 1766 (19,  $[M + Na]^+$ ).

Oleyl6) Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2-a]pyridine-2-phosphonate (1). A soln. of 20 (12 mg, 17.5  $\mu$ mol) in MeOH (1 ml) was treated at 25° with a 0.5m soln. of MeONa in MeOH (41 ml), and stirred for 45 min. MeOH (3 ml) was added, and the mixture was neutralized with Amberlite IRC50 (H<sup>+</sup> form). The resin was filtered and the filtrate evaporated. The residue (10 mg) was taken up in hot MeOH (ca. 0.25 ml), and 1 was precipitated at  $0^{\circ}$  by the addition of cold MeCN, separated by filtration, and dried under vacuum, affording  $1(8.5 \text{ mg}, 95\%)$ . White solid.  $R_f$  (AcOEt/MeOH/H<sub>2</sub>O 10:4:1) 0.2. IR (KBr): 3356s (br.), 2924s, 2853s, 1522m, 1466m, 1184m, 1066s, 828w. <sup>1</sup> H-NMR (CD3OD, 500 MHz): 0.89  $(t, J = 6.9, \text{ Me})$ ; 1.20 - 1.40  $(m, 22 \text{ H})$ ; 1.53 - 1.61  $(m, CH_2CH_2O)$ ; 1.98 - 2.07  $(m, CH_2CH=CHCH_2)$ ; 3.70  $(dd, J = 8.1, 9.5$ , irrad. at  $4.54 \rightarrow$  change, H $-C(7)$ ); 3.79 (q,  $J = 6.5$ , CH<sub>2</sub>CH<sub>2</sub>O); 3.84 (dd,  $J = 8.0, 9.5$ , irrad. at  $3.70 \rightarrow$  change, H $-C(6)$ );  $3.89 - 3.94$  (m, H $-C(5)$ ); 3.96 (dd, J = 4.0, 11.8, CH $-C(5)$ ); 4.18 (dd, J = 2.0, 11.8,  $CH'-C(5)$ ; 4.54 (d, J = 8.0, H – C(8)); 5.29 – 5.37 (m, CH<sub>2</sub>CH=CHCH<sub>2</sub>); 7.60 (d, J = 0.9, H – C(3)). <sup>13</sup>C-NMR (CD<sub>3</sub>OD, 125 MHz): 14.47 (q, Me); 23.77–30.94 (several d); 32.04 (dt, <sup>3</sup>J(C,P) = 7.3, CH<sub>2</sub>CH<sub>2</sub>O); 33.09  $(t, CH_2)$ ; 61.19  $(t, CH_2-C(5))$ ; 63.24  $(dt, {}^2J(C, P) = 5.5$ ,  $CH_2CH_2O)$ ; 65.88  $(d, C(5))$ ; 69.27, 69.59, 71.28  $(3d, C(6), C(7), C(8))$ ; 125.10  $(dd, {^2J}(C,P) = 29.1, C(3))$ ; 130.86, 130.92 (2d, CH=CH); 136.78  $(d, {^1J}(C,P) =$ 197.5, C(2)); 149.44  $(d, {}^{3}J(C, P) = 16.8, C(8a))$ . <sup>31</sup>P-NMR (CD<sub>3</sub>OD, 121 MHz): 5.89. FAB-MS: 553 (69, [M +  $\text{Na}^{+}$ ), 575 (100,  $[M - 1 + \text{Na}^{+}]$ ).

Phytanyl<sup>6</sup>) Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2a/pyridine-2-phosphonate (2). A soln. of 21 (15 mg, 20.6 µmol) in MeOH (1.2 ml) was treated with 0.5m MeONa in MeOH  $(49 \text{ µ})$  and stirred for 90 min. After dilution with MeOH  $(4 \text{ ml})$ , the mixture was neutralized with Amberlite IRC50 (H<sup>+</sup> form). The resin was filtered and the filtrate evaporated. The residue (13 mg) was taken up in hot MeOH (ca. 0.2 ml), and 2 was precipitated at  $0^{\circ}$  by the addition of cold MeCN. Filtration and drying under vacuum gave  $2(7 \text{ mg}, 61\%)$ . White solid.  $R_f$  (CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 3:1:0.1) 0.16. IR (KBr): 3382s (br), 2953s, 2912s, 2839s, 1522w, 1460m, 1375w, 1262w, 1189m, 1138m, 1065s, 901w, 856w, 805w.  $1H\text{-NMR (CD<sub>3</sub>OD, 300 MHz): 0.85 (m, Me); 0.89 (br. d, J = 7.2, irrad. at 1.53 \rightarrow change, 4 Me); 1.0-1.44$  $(m, 21 H)$ ; 1.53 (sept.,  $J = 8.8$ , irrad. at 0.87  $\rightarrow$  change, 1 H); 1.50 - 1.70  $(m, 3 H)$ ; 3.72  $(t, J = 8.8$ , irrad. at 4.19  $\rightarrow$  $d, J \approx 9.3$ , H $-C(7)$ ); 3.84 (m, irrad. at 3.96  $\rightarrow$  change, H $-C(6)$ , CH<sub>2</sub>O); 3.96 (m, CH $-C(5)$ , H $-C(5)$ ); 4.19 (dd, J = 4.1, 13.5, irrad. at 3.96  $\rightarrow$  br. s, CH' $-C(5)$ ); 4.58 (d, J = 8.1, H $-C(8)$ ); 7.66 (d, J = 1.2, H $-C($  $^{13}$ C-NMR (CD<sub>3</sub>OD, 75 MHz): 19.73 – 23.05 (several q); 25.46, 25.85 (2t); 29.12 – 33.95 (several d); 38.37 – 40.52 (several t); 61.04 (t, CH<sub>2</sub>-C(5)); 64.30 (dt, <sup>3</sup>J(C,P) = 6.1, CH<sub>2</sub>O); 61.04 (d, C(5)); 69.15, 69.25, 76.05 (3d, C(6),  $C(7)$ ,  $C(8)$ ); 125.47  $(dd, <sup>2</sup>J(C,P) = 26.8$ ,  $C(3)$ ); 135.54  $(d, <sup>1</sup>J(C,P) = 216.1$ ,  $C(2)$ ); 149.6  $(d, <sup>3</sup>J(C,P) = 13.4$ , C(8a)). <sup>31</sup>P-NMR (CD<sub>3</sub>OD): 4.30. FAB-MS: 281 (43,  $[(M - \frac{phytanylov y + 1}{)}$ , (561 (43,  $[(M + 1]^+)$ , 583  $(100, [M + Na]^+)$ , 605 (60,  $[M - 1 + 2 Na]^+$ ), 1188 (4,  $[2 M - 1 + 3 Na]^+$ ). HR-FAB-MS: 561.3676, (MH<sup>+</sup>; calc. 561.3668).

Dolichyl<sup>6</sup>) Hydrogen (5R,6R,7S,8S)-5,6,7,8-Tetrahydro-6,7,8-trihydroxy-5-(hydroxymethyl)imidazo[1,2a/pyridine-2-phosphonate (3). A soln. of 22 (14 mg, 8.02 µmol) in THF/MeOH 1:2 (0.75 ml) was treated with 0.4m of MeONa in MeOH (24  $\mu$ ) at 25° and stirred for 105 min. The soln. was diluted with THF (ca. 2 ml), neutralized with Amberlite IRC50 (H<sup>+</sup> form). The resin was filtered off and the filtrate evaporated. FC (CHCl<sub>3</sub>/ MeOH/H<sub>2</sub>O  $10:1:0.15 \rightarrow 10:2:0.25 \rightarrow 10:3.5:0.5$ , evaporation, washing with MeOH, and drying in vacuo gave 3 (9.3 mg, 75%).  $R_f$  (CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O 10:2:0.25) 0.28. IR (CCL<sub>4</sub>): 3286m (br.), 2960s, 2926s, 2855s, 1736w, 1598w, 1451m, 1376m, 1260w, 1194w, 1072m. <sup>1</sup>H-NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD/D<sub>2</sub>O 10:7:1, 500 MHz): 0.59  $(d, J = 6.5, \text{Me})$ ; 0.75 - 1.20  $(m, 15 \text{ H}$  instead of the expected 5 H); 1.40, 1.36, 1.34 (3s, 4 Me); 1.42 (s, 15 Me); 1.86  $-1.69$  (m, 70 H); 3.48 (br. t,  $J \approx 8.5$ , 1 H); 3.51  $-3.68$  (m, 4 H); 3.74 (br. d,  $J \approx 12$ , CH $-C(5)$ ); 3.91  $(br. d, J \approx 12, CH'-C(5))$ ; 4.80 – 4.95  $(m, 18 H)$ ; 7.23  $(br. s, H-C(3))$ . <sup>13</sup>C-NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD/D<sub>2</sub>O 10 : 7 : 1, 125 MHz): some signals are hidden by the noise; 15.36 – 18.36 (several q); 22.74, 22.86 (2q, 2 Me) 24.64 – 26.26 (several t); 28.82 (d, CH); 29.11 – 39.58 (several t); 60.78 (t, C(5)); 63.0 (t, CH<sub>2</sub> – C(5)); 64.5 (t, CH<sub>2</sub>O); 67.06, 67.5, 74.21 (3d, C(6), C(7), C(8)); 123.75 - 125.34 (several d); 130.71 - 134.84 (several s). <sup>31</sup>P-NMR (CDCl<sub>3</sub>/ CD3OD/D2O 10 : 7 : 1, 202 MHz) 7.4. FAB-MS: 1577 (100, [M 1]). MALDI-MS (THA (1,2,3,4-tetrahydroacridin-9-amine hydrochloride)/citrate 2:1): 1576.7  $(M^+)$ , 1599.0  $([M + Na]^+)$ . ESI-FT/MS/MS: 281 (100,  $[M - \text{dolichyloxy} + 2]^+$ , 1577 (84,  $[M + 1]^+$ ). HR-ESI-FT/MS: 1576.2602 (MH<sup>+</sup>; calc. 1576.2588).

## **REFERENCES**

- [1] J. A. Campbell, G. J. Davies, V. Bulone, B. Henrissat, Biochem. J. 1997, 326, 929.
- [2] S. Charnok, G. J. Davies, Biochemistry 1999, 38, 6369.
- [3] A. Vrielink, W. Rüger, H. P. C. Driessen, P. S. Freemont, *EMBO* 1994, 13, 3413.
- [4] E. H. Holmes, Z. Xu, A. L. Sherwood, B. A. Macher, J. Biol. Chem. 1995, 270, 8145.
- [5] B. W. Murray, S. Takayama, J. Schultz, C.-W. Wong, Biochemistry 1996, 35, 11183.
- [6] C. J. Britten, M. Bird, Biochim. Biophys. Acta 1997, 1334, 57.
- [7] S. C. Kim, A. N. Singh, F. M. Raushel, Arch. Biochem. Biophys. 1988, 267, 54.
- [8] B. W. Murray, V. Wittmann, M. D. Burkart, S.-C. Hung, C.-H. Wong, *Biochemistry* 1997, 36, 823.
- [9] Y. Ichikawa, Y.-C. Lin, D. Dumas, G.-J. Shen, E. Garcia-Juncade, M. A. Williams, R. Bayer, C. Ketcham, L. E. Walker, J. C. Paulson, C.-H. Wong, J. Am. Chem. Soc. 1992, 114, 9283.
- [10] T. Hayashi, B. W. Murray, R. Wang, C.-H. Wong, Bioorg. Med. Chem. 1997, 5, 497.
- [11] L. Qiao, B. W. Murray, M. Shimazaki, J. Schultz, C. H. Wong, J. Am. Chem. Soc. 1996, 118, 7653.
- [12] I. Jefferies, B. R. Bowen, *Bioorg. Med. Chem. Lett.* **1997**, 7, 1171.
- [13] S. Cai, M. R. Stroud, S. Hakomori, T. Toyokuni, J. Org. Chem. 1992, 57, 6693.
- [14] R. R. Schmidt, K. Frische, Bioorg. Med. Chem. Lett. 1993, 3, 1747.
- [15] R. D. Cumming, in 'Glycoconjugates, Composition, Structure, and Function', Eds. H. J. Allen and E. C. Kisailus, Marcel Dekker, New York, 1992, p. 333.
- [16] N. H. Behrens, L. F. Leloir, Proc. Natl. Acad. Sci. U.S.A. 1970, 66, 153.
- [17] C. J. Waechter, Ann. Rev. Biochem. **1976**, 45, 95.
- [18] P. Burda, M. Aebi, *Glycobiology* **1998**, 8, 455.
- [19] T. Granier, N. Panday, A. Vasella, *Helv. Chim. Acta* 1997, 80, 979.
- [20] K. Tatsuta, S. Miura, S. Ohta, H. Gunji, *Tetrahedron Lett*. **1995**, 36, 1085.
- [21] K. Tatsuta, S. Miura, H. Gunji, Bull. Chem. Soc. Jpn. 1997, 70, 427.
- [22] T. D. Heightman, M. Locatelli, A. Vasella, Helv. Chim. Acta 1996, 79, 2190.
- [23] N. Panday, A. Vasella, in preparation.
- [24] K. Tatsuta, S. Miura, Tetrahedron Lett. 1995, 36, 6721.
- [25] T. Hirao, T. Masunaga, Y. Ohshiro, T. Agawa, Synthesis **1981**, 56.
- [26] T. Hirao, T. Masunaga, N. Yamada, Y. Ohshiro, T. Agawa, Bull. Chem. Soc. Jpn. 1982, 55, 909.
- [27] Y. Xu, J. Zhang, Synthesis 1984, 778.
- [28] H. Lei, M. S. Stoakes, A. W. Schwabacher, Synthesis 1992, 1255.
- [29] J. Lin, C. M. Thompson, *J. Heterocycl. Chem.* **1994**, 31, 1701.
- [30] N. Panday, Ph. D. Thesis, in preparation.
- [31] K. Sonogashira, Y. Tohda, N. Hagihara, Tetrahedron Lett. 1975, 50, 4467.
- [32] C. Cai, A. Vasella, *Helv. Chim. Acta* 1995, 78, 2053.

- [33] T. Morita, Y. Okamoto, H. Sakurai, Bull. Chem. Soc. Jpn. 1978, 51, 2169.
- [34] D. Redmore, *J. Org. Chem.* **1973**, 38, 1306.
- [35] L. A. Vargas, L. X. Miao, A. F. Rosenthal, Biochim. Biophys. Acta 1984, 796, 123.
- [36] L. Qiao, J. C. Vederas, J. Org. Chem. 1993, 58, 3480.
- [37] J. Coste, M.-N. Dufour, A. Pantaloni, B. Castro, Tetrahedron Lett. 1990, 31, 669.
- [38] N. Galeotti, J. Coste, P. Bedos, P. Jouin, Tetrahedron Lett. 1996, 37, 3997.
- [39] K. Yamauchi, F. Une, S. Tabata, M. Kinoshita, J. Chem. Soc., Perkin Trans. 1 1986, 765.
- [40] D. V. Patel, K. Rielly-Gauvin, D. E. Ryono, Tetrahedron Lett. 1990, 31, 5591.
- [41] O. Mitsunobu, M. Eguchi, *Bull. Chem. Soc. Jpn.* **1971**, 44, 3427.
- [42] O. Mitsunobu, Synthesis **1981**, 1.
- [43] R. Hoos, A. B. Naughton, A. Vasella, *Helv. Chim. Acta* 1992, 75, 1802.
- [44] Y. Sakata, Y. Hirano, H. Tatemitsu, S. Misumi, H. Ochiai, H. Shibata, Tetrahedron 1989, 45, 4717.

Received April 30, 1999